A General Synthesis of N-Glycosides. IV.¹ Synthesis of Nucleosides of Hydroxy and Mercapto N-Heterocycles

U. Niedballa and H. Vorbrüggen*

Research Laboratories of Schering A. G., Department Allgemeine Synthetische Chemie, 1 Berlin 65, Muellerstrasse 170–172, Germany

Received March 29, 1974

A variety of silylated hydroxy and mercapto N-heterocycles react with 1-O-acetyl-2,3,5-tri-O-benzoyl- β -D-ri-bofuranose in the presence of Friedel-Crafts catalysts to give N-ribosylated heterocycles in generally good yields. In the case of silylated pyrimidin-4-one both possible N-ribosides are obtained.

The smooth reaction of silylated uracils with 1-O-acylated mono- 2 and oligosaccharides 1 in the presence of $SnCl_4$ to give the corresponding nucleosides in good to excellent yields induced us to investigate analogous reactions of silylated hydroxy and mercapto N-heterocycles.

On reaction of silylated pyridin-2-one (1) with 1-O-acetyl-2,3,5-tri-O-benzoyl- β -D-ribofuranose (2) in the presence of SnCl₄ we obtained the crystalline benzoylated nucleoside 3 in 85% yield, which had been prepared previously by Pischel and Wagner³ via the O-glycoside followed by

O,N rearrangement. Even in the presence of a large excess of $\mathrm{SnCl_4}$ and longer reaction times at room temperature silylated pyridin-4-one (4) gave with 2 in either 1,2-dichloroethane or acetonitrile the expected benzoylated nucleoside 5a and the crystalline O-glycoside 6a (both products in $\sim 10\%$ yield). On refluxing the reaction mixture for 1 hr in 1,2-dichloroethane, only 2% O-riboside 6a was still present and 63% nucleoside 5a was obtained, which gave on methanolysis the crystalline free nucleoside 5b. 5a had been pre-

^a After methanolysis.

pared earlier by Pischel and Wagner³ by a Hilbert–Johnson reaction of 4-ethoxypyridine as well as by rearrangement of 6a with HgBr₂ in boiling toluene. Our experiments to rearrange 6a with SnCl₄ in boiling 1,2-dichloroethane as well as

acetonitrile gave mostly decomposition products and only low yields of 5a.

These results support the hypothesis that the N-glycosides are usually obtained directly via the N-quaternary salt from the silylated heterocycle and the sugar component in the presence of $SnCl_4$ and that a side reaction leads to the formation of O- or S-glycosides, which are then partially rearranged to the N-glycosides or decomposed by the catalyst.

Silylated pyrimidin-2-one (7a) afforded 8a in 73% yield, which had been prepared earlier by the mercuric salt^{4,5} method.

The silylated 2-mercaptopyrimidine (7b) gave analogously the 2-thio nucleoside 8b in 97% yield, which was readily identified by the typical downfield shift of the H-1′ proton ($\delta_{\text{H-1'}}$ 7.02) compared to 8a (6.35) of 1 ppm.⁶ 8b was recently prepared by Wightman and Holý⁷ via the S-riboside and subsequent rearrangement with SnCl₄ in acetonitrile.

To gain further insight into the mode of electrophilic attack of sugar cations on silylated hydroxy N-heterocycles we treated silylated pyrimidin-4-one (9) with 2 in the presence of SnCl₄ and obtained beside the expected crystalline ortho quinoid nucleoside 10a (attack A) also the crystalline para quinoid product 11a (attack B). Depending on the polarity of the solvent⁸ we isolated 26% 10a and 60% 11a in 1,2-dichloroethane and 27% 10a and 38% 11a in acetonityila

The formation of 11a was surprising since methylation of 4-hydroxypyrimidine had yielded >50% 1,6-dihydro-1-methylpyrimidin-6-one (corresponding to 10) and \sim 20% 4-methoxypyrimidine but apparently no 1,4-dihydro-1-methylpyrimidin-4-one.⁹ Furthermore, the mercuric salt^{5,10} as well as the fusion method¹¹ had apparently afforded only the ortho quinoid nucleoside 10a.

The structures of 11a as well as 11b, which had been prepared previously by Raney nickel desulfurization of 2-thiouridine, 12 can be deduced from their typical uv 11b [(CH₃OH), λ_{max} 243 nm (ϵ 15,290)]; 1,4-dihydro-1-methylpyrimidin-4-one (H₂O), λ_{max} 240 nm (ϵ 14,640)⁹] as well as nmr data. 13

^a After methanolysis.

The preferential formation of 11a in the silyl Hilbert-Johnson reaction in 1,2-dichloroethane might be due to the steric hindrance of the N-3 nitrogen by the bulky trimethylsilyloxy group, favoring attack B on 9.

Thus the preponderance of the ortho quinoid 12a over the para quinoid form 13a as proved by uv, ir, ¹⁴ as well as nmr evidence ¹³ does not necessarily have a bearing on the reactions of the silylated form 9. Furthermore, Pitha ¹⁵ demonstrated by uv measurements, in the case of 2-ethoxypyrimidin-4-one (13b), that the ortho quinoid form 12b predominates over 13b only in chloroform, while in water both the ortho quinoid 12b and the para quinoid form 13b are present in about equal amount. Thus the energy difference between the forms 12 and 13 is probably very low.

Since 6-azaisocytosine occurs mainly in the para quinoid form 15 and not in the ortho quinoid form 14,¹⁶ we wondered how as-triazin-5-one (5-hydroxy-as-triazine) would behave in the silyl Hilbert-Johnson reaction.

$$\begin{array}{c}
0 \\
HN \\
N
\end{array} \longrightarrow \begin{array}{c}
0 \\
N \\
N
\end{array} \longrightarrow \begin{array}{c}
N \\
N \\
N
\end{array}$$
14

Uv data as well as alkylation studies had established for as-triazin-5-one (5-hydroxy-as-triazine)¹⁷ that both possible tautomers, **16a** and **17a**, are present. On methylation with diazomethane in methanol as well as with methyl iodide and sodium methoxide the corresponding N-methyl derivatives **16b** and **17b** were obtained in roughly equal yields.¹⁷

In contrast to the methylation studies, silylated astriazin-5-one (18) afforded on reaction with 2 and SnCl₄ in

1,2-dichloroethane as well as acetonitrile almost exclusively the crystalline para quinoid nucleoside 19a in 67% yield and apparently only traces of the expected ortho quinoid 20a, which was not isolated.

^a After methanolysis.

The structure of 19a and the free nucleoside 19b is supported by nmr data [19a (H-1'), δ 5.96; 19b (H-1'), δ 5.79] and especially by the uv spectrum of 19b in methanol [λ_{max} 242 nm (ϵ 13,600), 269 (sh, 5240)], which agrees closely with the uv spectrum of the methyl derivative 17b (EtOH) [λ_{max} 242 nm (ϵ 11,400), 260 (sh, 4670)¹⁷].

 \mathbf{b} , a $\mathbf{R} = \mathbf{H}$

The para quinoid nucleoside 19a was also prepared by Raney nickel desulfurization of 2',3',5'-tri-O-benzoyl- β -D-2-thio-6-azauridine (21), which gave, besides 19a, the corresponding crystalline 1,2-dihydro nucleoside 22 in about equal yield. Oxidation of 22 with tert- butyl hypochlorite—

$$\begin{array}{c} & & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$$

triethylamine¹⁸ gave 60% **19a** as well as $2-(2,3,5-\text{tri}-O-\text{ben-zoyl}-\beta-D-\text{ribofuranosyl})-6-\text{chloro-}2,3,4,5-\text{tetrahydro-}1,2,4-\text{triazin-}5-\text{one}$ **(23)**.

Analogously, the silylated quinolin-2-one (24) was converted into the nucleoside 25a in 79% yield. Silylated 7-methoxyquinolin-4-one-3-carboxylic acid ethyl ester (26) afforded the nucleoside 27a in about 82% yield.

^a After methanolysis. ^b After saponification.

These results demonstrate that a wide variety of silylated ortho or para quinoid hydroxy or mercapto N-heterocycles react with 2 and in all probability with other suitable sugar derivatives² in the presence of SnCl₄ to give the corresponding N-glycosides in good to excellent yields.

Following our preliminary publication, 2 a number of different groups have successfully used our procedure for the synthesis of nucleosides of hydroxy N-heterocycles. $^{19-25}$

Experimental Section

For instruments, adsorbents, and the purification of solvents compare part I.² Tlc systems: system A [ethyl acetate-methanol (9:1)]; system B (ethyl acetate); and system C [n-butyl acetate-methyl glycol- H_2O (4:1:2)].²⁶

The heterocyclic bases were purchased from Fluka AG. The astriazin-5-one was prepared according to Lee and Paudler. The 3-carbethoxy-7-methoxy-1,4-dihydroquinolin-4-one was a gift from Dr. Albrecht, Schering A. G. The heterocyclic bases were sily-lated in high yields according to standard methods. 2

1-(2,3,5-Tri-O- benzoyl- β -D-ribofuranosyl)-1,2-dihydropyridin-2-one (3). To 2 (9.1 g, 18 mmol) and 2-trimethylsilyloxypyridine (1, 22.6 mmol) in 1,2-dichloroethane (200 ml) SnCl₄ (2.9 ml, 24.8 mmol) in 1,2-dichloroethane (50 ml) was added. After stirring overnight and usual work-up² crystallization (CCl₄) gave 3 in long needles; yield, 8.3 g (85.2%); mp 139-140° (lit. 3 139-142°); $[\alpha]^{20}$ D 61° (c 1, CHCl₃); nmr (CDCl₃) 5 6.59 (d, 1, $J \approx 4$ Hz, H-1')

61° (c 1, CHCl₃); nmr (CDCl₃) δ 6.59 (d, 1, J \approx 4 Hz, H-1′) Anal. Calcd for C₃₁H₂₅NO₈ (539.55): C, 69.01; H, 4.67; N, 2.60. Found: C, 68.99; H, 4.84; N, 2.73.

 $1\hbox{-}(2,\!3,\!5\hbox{-}\mathrm{Tri-}O\hbox{-}\ benzoyl\hbox{-}\beta\hbox{-}D\hbox{-}\mathrm{ribofuranosyl})\hbox{-}1,\!4\hbox{-}\mathrm{dihydropyr}\hbox{-}$

idin-4-one (5a), 4-[(2,3,5-Tri-O-benzoyl- β -D-ribofuranosyl)-oxy]pyridine (6a). To 2 (4.5 g, 8.9 mmol) and 4 (1.7 g, 10 mmol) in 1,2-dichloroethane (150 ml) SnCl₄ (1.2 ml, 10.25 mmol) in 1,2-dichloroethane (50 ml) was added and the mixture was refluxed for 1 hr. After work-up, the residue was chromatographed on silica gel (250 g). Elution with ethyl acetate gave the crystalline O-riboside 6a (ethanol) (tlc, system A): yield, 165 mg (2.3%); mp 139–140° (lit.³ 139–142°); nmr (CDCl₃) δ 6.1–5.9 (m, 3, H-1′, H-2′, H-3′).

Anal. Calcd for $C_{31}H_{25}NO_8$ (539.55): C, 69.01; H, 4.67; N, 2.60. Found: C, 68.75; H, 4.78; N, 2.54.

Ethyl acetate–methanol (9:1) eluted the N-riboside **5a**: yield; 3.227 g (63%); amorphous; nmr (CDCl₃) δ 5.9–5.6 (m, 3, H-1′, H-2′, H-3′); $[\alpha]^{23}$ D -150.5° (c 1, CHCl₃)

Anal. Calcd for $C_{31}H_{25}NO_8$ (539.55): C, 69.01; H, 4.67; N, 2.60. Found: C, 68.83; H, 4.79; N, 2.52.

1-(β -D-Ribofuranosyl)-1,4-dihydropyridin-4-one (5b). 5a (1.5 g, 2.8 mmol) was kept in dry methanolic ammonia (60 ml) overnight. After evaporation in vacuo and partition between water and ether, the aqueous layer was concentrated in vacuo to a viscous oil, which crystallized (ethanol) in colorless prisms: yield of 5b, 514 mg (81.3%); mp 128-130°; [α]²³D -89.6° (c 1, ethanol + H_2O = 2:1); nmr (D_2O) δ 5.57 (d, 1, J = 3.5 Hz, H-1'); uv (CH₃OH) $\lambda_{\rm max}$ 266 nm (18,420).

Anal. Calcd for $C_{10}H_{13}NO_5$ (227.22): C, 52.86; H, 5.77; N, 6.17. Found: C, 52.81; H, 5.89; N, 6.08.

1-(2,3,5-Tri-O- benzoyl- β -D-ribofuranosyl)-1,2-dihydropyrimidin-2-one (8a). To 2 (5.045 g, 10 mmol) and 7a (13.4 mmol) in CH₃CN (100 ml) SnCl₄ (1.73 g, 14.1 mmol) in CH₃CN (50 ml) was added. After 22° and work-up the residue (5.5 g) was chromatographed on silica gel (250 g) using ethyl acetate—methanol (9:1) as eluent to give 8a as a white amorphous powder after precipitation from n- hexane: mp 139–142° [after precipitation from ethanol, mp 155–158° (lit.⁶ mp 154–158°)]: yield; 3.89 g (72.9%); nmr (CDCl₃) δ 6.35 (d, 1, J = 3.5 Hz, H-1′).

Anal. Calcd for $C_{30}H_{24}N_2O_8$ (540.54): C, 66.66; H, 4.48; N, 5.18. Found; C, 66.59; H, 4.52; N, 5.19.

1-(2,3,5-Tri-O- benzoyl- β -D-ribofuranosyl)-1,2-dihydropy-rimidine-2-thione (8b). To 2 (5.045 g, 10 mmol) and 7b in CH₃CN (100 ml) SnCl₄ (1.73 ml, 14.1 mmol) in CH₃CN (100 ml) was added. According to tlc (system B) the reaction was complete after 5 min at 22° and was worked up after an additional 5 min. The residue was dissolved in ethyl acetate (50 ml) and added slowly to n- hexane (2 l.). The precipitate, a yellow powder (5.23 g), was homogenous according to tlc (system B). The hexane solution was evaporated to dryness and the residue (0.49 g) was dissolved in ethyl acetate and again precipitated by n-hexane to give a second crop (0.18 g): total yield of 8b, 5.41g (96.7%); amorphous; $[\alpha]^{23}$ D 189.7° (c 1, CHCl₃); nmr (CDCl₃) δ 7.02 (d, 1, J = 2 Hz, H-1').

Anal. Calcd for C₃₀H₂₄N₂O₇S (559.63): C, 64.39; H, 4.86; N, 5.01; S, 5.73. Found: C, 64.31; H, 4.97; N, 5.03; S, 5.66.

1-(2,3,5-Tri-O- benzoyl- β -D-ribofuranosyl)-1,6-dihydropyrimidin-6-one (10a) and 1-(2,3,5-Tri-O- benzoyl- β -D-ribofuranosyl)-1,4-dihydropyrimidin-4-one (11a). A. Reaction in 1,2-Dichloroethane. To 2 (4.14 g, 8.20 mmol) and 9 (10 mmol) in 1,2-dichloroethane (100 ml) SnCl₄ (1.60 ml, 13.67 mmol) in 1,2-dichloroethane (50 ml) was added. After 0.5 hr at 15°, 3.5 hr at 22°, and work-up, tlc (system A) showed the formation of two products. The residue was dissolved in hot ethyl acetate (~150 ml) from which 11a crystallized in long needles: yield, 2.66 g (60%); mp 224–226°; [α]²³D –138.9° (c 1, CHCl₃); nmr (CDCl₃) δ 5.9–5.6 (m, 3, H-1′, H-2′, H-3′).

Anal. Calcd for C₃₀H₂₄N₂O₈ (540.39): C, 66.66; H, 4.48; N, 5.18. Found: C. 66.61; H, 4.59; N, 5.28.

11a (1.4 g, 2.6 mmol) gave with methanolic ammonia (16 hr at 22°) 11b (521 mg, 87.8%): mp 122–124° (ethanol–H₂O); $[\alpha]^{23}$ D –75.8° [c 1, ethanol–H₂O (2:1)]; nmr (D₂O) δ 5.67 (d, 1, J = 5 Hz, H-1′).

Anal. Calcd for $C_9H_{12}N_2O_5$ (228.21): C, 47.37; H, 5.30; N, 12.28. Found: C, 47.45; H, 5.67; N, 12.15.

The mother liquor of 11a was concentrated in vacuo and poured into n-hexane to remove sugar derivatives. The precipitate crystallized (ethyl acetate–n-hexane) to give 10a in needles: yield, 1.16 g (26.4%); mp (128–132°, changing point) 157–158° (lit. 5 157–157.5°); nmr (CDCl₃) δ 6.24 (d, 1, J = 3 Hz, H-1′).

B. Reaction in CH₃CN. To 2 (5.045 g, 10 mmol) and 9 (10 mmol) in CH₃CN (100 ml) SnCl₄ (2.0 ml, 17.1 mmol) in CH₃CN (100 ml) was added. After 0.5 hr at 15° and 3.5 hr at 22° work-up gave 27% 10a and 38% 11a.

 $2-(2,3,5-Tri-O-benzoyl-\beta-D-ribofuranosyl)-2,5-dihydro-1,2,4-triazin-5-one (19a). To a solution of 2 and 18 (12 mmol) in$

CH₃CN (60 ml) SnCl₄ (1 ml, 8.16 mmol) in CH₃CN (40 ml) was added. After 24 hr at 22° and work-up the crude product (5.3 g) was chromatographed on silica gel (250 g) using n-hexane-ethyl acetate (1:9) as eluent. The crude nucleoside was dissolved in ethyl acetate and poured into pentane. The precipitate crystallized (ethanol) to give 19a as needles (3.61 g, 66.6%); mp 117-119°; $[\alpha]^{23}$ D -94° (c 1, CHCl₃); nmr (CDCl₃) δ 8.53 (d, 1, J = 1.5 Hz, H-3), 5.96 (d, 1, J = 3.5 Hz, H-1′).

Anal. Calcd for C₂₉H₂₃N₃O₈ (541.53): C, 64.32); H, 4.28; N, 7.76. Found: C, 64.25; H, 4.41; N, 7.68.

2-(β-D-Ribofuranosyl)-2,5-dihydro-1,2,4-triazin-5-one (19b). 19a (2.5 g, 4.62 mmol) in dry methanolic ammonia (60 ml) was stored overnight to give 19b in needles (2-propanol-H2O): yield, 876 mg (83.5%); mp 116–119°; $[\alpha]^{23}$ D -74.2° [c 1, ethanol– H_2 O) (2:1)] mmr (D₂O) δ 8.93 (d, 1, $J \approx 1$ Hz, H-3), 8.01 (d, 1, $J \approx 1$ 1 Hz, H-6), 5.79 (d, 1, J = 3.5 Hz, H-1'.

Anal. Calcd for $C_8H_{11}N_3O_5$ (229.20): C, 41.92; H, 4.84; N, 18.34. Found: C, 41.95; H, 5.12; N, 18.19.

Desulfurization of 2-Thio-6-azauridine 2'.3'.5'-Tri-O-benzoate (21) to 19a and 2-(2,3,5-Tri-O-benzoyl-β-D-ribofuranosyl)-2,3,4,5-tetrahydro-1,2,4-triazin-5-one (22). 21 (2.869 g, 5 mmol) and benzoic acid (0.61 g, 5 mmol) were suspended, stirred in abs ethanol (150 ml), and cooled to 0°, and freshly prepared W-2 Raney nickel²⁷ (12 g) was added, which had been washed with ethanol. After 4, 8, and 12 hr further portions of Raney nickel (10 g each) were added at 0° until the starting material had nearly disappeared according to tlc (system C). Thiourea (2 g) was added and stirring was continued for 1 hr at 0°. After filtering the excess Raney nickel, washing with cold ethanol (500 ml), and extracting the reagent with boiling ethanol (1 l.) the filtrates were evaporated and the residue was taken up in chloroform (200 ml). The chloroform solution was extracted with saturated NaHCO3 solution (150 ml), dried (MgSO₄), and evaporated to give a crude product (1.828 g), which consisted according to tlc (system C) of an ~ 1:1 mixture of 19a and the slightly slower moving 22. On standing in ethanol (30 ml) 22 crystallized: yield, 0.82 g (30%); mp 193–197°; $[\alpha]^{23}$ D 4.4° (c 1, CHCl₃); nmr (ČDCl₃ + D_2O) δ 6.75 (s, 1, H-6), 5.34 (d, 1, $J=3.5~{\rm Hz},\,{\rm H-1'});\,{\rm uv}\,\,({\rm CH_3OH})\,\,\lambda_{\rm max}\,\,({\rm pH}\,\,7)\,\,281~{\rm nm}\,\,)\epsilon\,\,5340),\,\lambda_{\rm max}$ (pH 13) 301 (3490), λ_{max} (pH 1) 281 (5350).

Anal. Calcd for C₂₉H₂₅N₃O₈ (543.51): C, 64.08; H, 4.64; N, 7.73. Found: C, 63.65; H, 4.74; N, 8.13. An aliquot (0.2 g) of the evaporated mother liquor (0.98 g) was separated by preparative tlc on silica gel (system C) to give pure 19a (83 mg, 16%), which was identified by tlc and by comparison of melting point and ir spectrum with authentic 19a.

Conversion of 22 to 19a and 2-(2,3,5-Tri-O-benzoyl-β-D-ribofuranosyl)-6-chloro-2,3,4,5-tetrahydro-1,2,4-triazin-5-one (23). A solution of 22 (543 mg, 1 mmol) in chloroform (20 ml) and triethylamine (0.14 ml, 1 mmol) was stirred at -10° and tert-butyl hypochlorite was added with vigorous stirring until 22 had disappeared according to tlc (system C). The yellow solution was washed with sodium thiosulfate solution, dried (Na₂SO₄), and evaporated to a light brown oil, which was chromatographed on silica gel (50 g). Elution with hexane-ethyl acetate (1:1) gave first 23 (112 mg, 19.3%) followed by 19a (316 mg, 58.4%), which was identified by tlc and ir comparison with an authentic sample of 19a.

23 was recrystallized (2-propanol): mp 173–175°; $[\alpha]^{23}D$ –18.8° (c 1, CHCl₃); nmr (CDCl₃) δ 5.30 (d, 1, J = 3.5 Hz, H-1'

Anal. Calcd for C₂₉H₂₄N₃O₈Cl (577.99): C, 60.26; H, 4.19; N, 7.27; Cl, 6.13. Found: C, 60.64; H, 4.08; N, 7.66; Cl, 6.18.

 $1-(2,3,5-\text{Tri-}O-\text{benzoyl-}\beta-D-\text{ribofuranosyl})-1,2-\text{dihydroqui-}$ nolin-2-one (25a). To 2 (5.045 g, 10 mmol) and 24 (12 mmol) in CH₃CN (100 ml) SnCl₄ (1.5 ml, 12.2 mmol) in CH₃CN (670 ml) was added. After 42 hr at 22° and work-up, the residue was treated with charcoal and chromatographed on silica gel (250 g) using nhexane-ethyl acetate (2:1) as eluent. 25a was obtained as a white foam: yield; 4.65 g (78.9%); amorphous; $[\alpha]^{23}$ D 4.7° (c 1, CHCl₃); nmr (CDCl₃) δ 6.77 (d, 1, J = 3 Hz, H-1'), 6.61 (d, 1, J = 9 Hz, H-

Anal. Calcd for C₃₅H₂₇NO₈ (589.61); C, 71.30; H, 4.62; N. 2.38. Found: C, 71.19; H, 4.90; N, 2.45.

25a (2.9 g, 4.92 mmol) gave with methanolic ammonia crystalline (ethanol) **25b** (1.20 g, 87.9%): mp 142–143°: $[\alpha]^{23}$ D –9.6° $[c\ 1]$, ethanol– H_2 O (2:1)]; nmr (pyridine- d_5) δ 8.09 (d, 1, J=9 Hz, H-4), 7.12 (d, 1, J = 5 Hz, H-1'), 6.57 (d, 1, J = 9 Hz, H-3).

Anal. Calcd for C₁₄H₁₅NO₅ (277.28): C, 60.64; H, 5.45; N, 5.05. Found: C, 60.57; H, 5.73; N, 4.99.

1-(2,3,5-Tri-O-benzoyl-β-D-ribofuranosyl)-3-carbethoxy-7-methoxy-1,4-dihydroquinolin-4-one (27a). To 2 (1.73 g, 3.43 mmol) and 26)1.2 g, 3.76 mmol) in 1,2-dichloroethane (60 ml) SnCl₄ (0.5 ml, 4.27 mmol) in 1,2-dichloroethane (10 ml) was added. After 2 hr at 22° and work-up the residue was crystallized (ethanol) to give 27a as needles (1.93 g, 81.5%): mp 133–136°: $[\alpha]^{20}$ D -80.4° (c 1, CHCl₃); nmr (CDCl₃) δ 8.94 (s, 1, H-2), 6.45 (d, 1, J = 4.5 Hz, H-1'), 3.75 (s, 3, Ar OH₃).

Anal. Calcd for C₃₉H₃₃NO₁₁ (691.70): C, 67.72; H, 4.81; N, 2.03. Found: C, 67.58; H, 4.97; N, 2.11.

27a (0.448 g, 0.695 mmol) in 0.5 N methanolic sodium hydroxide (30 ml) was stirred for 2 hr at 22° and passed through a column of Dowex 50 H⁺ (20 ml), which was washed with methanol-H₂O (3:1). The filtrate was evaporated and the residue was partitioned between water and ether. The aqueous solution was evaporated in vacuo and the residue crystallized (ethanol-H₂O) to give 27b: yield, 199 mg (81.6%); mp 213°.

Anal. Calcd for C₁₆H₁₇NO₈ (351.32): C, 54.70; H, 4.88; N, 3.99. Found: C, 54.82; H, 4.97; N, 3.86.

Acknowledgment. We are indebted to Dr. G.-A. Hover and Dr. D. Rosenberg for the nmr and mass spectra and Diplom-Ing. G. Huber for the analyses.

Registry No. 1, 18292-04-1; 2, 6974-32-9; 3, 5116-31-4; 4, 27248-04-0; 5a, 18342-24-0; 5b, 52554-33-3; 6a, 18342-23-9; 7a, 52523-22-5; 7b, 52523-23-6; 8a, 52523-24-7; 8b, 49625-10-7; 9, 52523-25-8; 10a, 5116-20-1; 11a, 52523-26-9; 11b, 21052-20-0; 18, 52523-27-0; 19a, 52523-28-1; 19b, 52554-34-4; 21, 27089-55-0; 22, 52523-29-2; 23, 52523-30-5; 24, 52523-31-6; 25a, 52523-32-7; 25b, 52523-33-8; **26**, 52523-34-9; **27a**, 35068-46-3; **27b**, 35068-65-6.

References and Notes

- (1) Synthesis of Nucleosides No. 12. Part III, J. Org. Chem., 39, 3664
- U. Niedballá and H. Vorbrüggen, *J. Org. Chem., 39*, 3654 (1974), part I of this series. Preliminary publication: U. Niedballa and H. Vorbrüggen,
- Angew. Chem., Int. Ed. Engl., 9, 461 (1970).
 H. Pischel and G. Wagner, Arch. Pharm. (Weinheim), 300, 602 (1967).
 R. Funakoshi, M. Irie, and T. Ukita, Chem. Pharm. Bull. (Tokyo), 9, 406
- H. Pischei, A. Holý, and G. Wagner, Collect. Czech. Chem. Commun., 37, 3475 (1972).
- (6) (a) H. Vorbrüggen, P. Strehlke, and G. Schulz, Angew. Chem., Int. Ed. Engl., 8, 976 (1969); (b) G. Cleve, G.-A. Hoyer, G. Schultz, and H. Vorbrüggen, Chem. Ber., 106, 3062 (1973).

 R. Wightman and A. Holý, Collect. Czech. Chem. Commun., 38, 1381
- (1973).
- For the influence of the polarity of the solvent on the mode of attack on silylated pyrimidines, compare U. Niedballa and H. Vorbrüggen, *J. Org. Chem.*, **39**, 3660 (1974), part II of this series.

 D. J. Brown, E. Hoerger, and S. F. Mason, *J. Chem. Soc.*, 211 (1955).

- M. Prystaš, *Collect. Czech. Chem. Commun.*, 33, 1813 (1968).
 W. Pfleiderer and R. K. Robins, *Chem. Ber.*, 98, 1511 (1965).
 H.-J. Lee and P. W. Wigler, *Biochemistry*, 7, 1427 (1968).
 L. Bauer, G. E. Wright, B. A. Mikrut, and C. L. Bell, *J. Heterocycl. Chem.*, 24, 7476. 2, 447 (1965). (14) A. R. Katritzky and L. M. Lagowski, *Advan. Heterocycl. Chem.*, 1, 368

- (150 J. Pitha, J. Org. Chem., 35, 903 (1970).
 (16) J. Pitha, P. Fiedler, and J. Gut, Collect. Czech. Chem. Commun., 31, 1864 (1966).
- 1864 (1966).
 (17) (a) J. Lee and W. W. Paudler, J. Heterocycl. Chem., **9**, 995 (1972); (b) D. J. Brown and R. L. Jones, Aust. J. Chem., **25**, 2711 (1972).
 (18) M. Julia and J. Y. Lallemand, Bull. Soc. Chim. Fr., 2058 (1973).
 (19) H. G. Gassen, FEBS Lett., **14**, 225 (1971).
 (20) M. Bobek and A. Bloch, J. Med. Chem., **15**, 164 (1972).
 (21) M. Bobek, A. Bloch, and S. Kuhar, Tetrahedron Lett., 3493 (1973).
 (22) T. Ling Chwang and C. Heidelberger, Tetrahedron Lett., 95 (1974).

- (22) H. Ling Chwang and C. Heidelberger, Ietrahedron Lett., 95 (1974).
 (23) H. Steinmaus, German Patent Application DOS 2224379.
 (24) W. Pfleiderer and M. Schrammer, Chem. Ber., 104, 1915 (1971).
 (25) M. Ott and W. Pfleiderer, Chem. Ber., 107, 339 (1974).
 (26) For the use of two-phase tic systems, compare H. Vorbrüggen and P. Strehlke, Chem. Ber., 106, 3039 (1973).
 (27) "Organic Syntheses," Collect. Vol. III, Wiley, New York, N. Y., 1955, p. 181